

# Determining the Ideal Growth Kinetics of Carbon Nanotubes

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#### Background

- Carbon nanotubes (CNTs) can carry 550,000 times more current per area than copper
- CNTs are extremely efficient conductors of electricity but are difficult to consistently produce in uniform samples with controlled diameter
- Historically made with iron, which is unstable, so higher melting point element that is stable is necessary
- Research project aimed to study optimal growth conditions
- Bimetal alloy nanoparticle consisting of high melting metal and an iron group metal in 4 sample sizes XS, S, M, and XL, whose diameter were 1.2-3 nm, were spin-coated on a silicon wafer and annealed with H<sub>2</sub> for 0, 10, 20, and 30 minutes and then studied for changes
- Particle evolution on the samples was then studied with an Atomic Force Microscope (AFM) and their nanotube density and diameters were recorded
- AFM uses a laser reflected off a cantilever to determine the height of the catalyst particles



Atomic Force Microscope



Semiconducting CNTs

### Sample Name: CAT 1215202201 XS Reduction Temperature: 900°C:



### Sample Name: CAT 0208202302 S Reduction Temperature: 900°C:



### Sample Name: CAT 1025202203 M1 Reduction Temperature: 900°C:



## Sample Name: CAT 1215202203 XL Reduction Temperature: 900°C:



# Chemical Vapor Deposition (CVD) Results for the Growth of Nanotubes for the M1 Silicon Wafer Catalyst



Note: The formula for calculating the diameter of a nanotube is  $d_{cnt}$ = 248 / raman shift

#### **Catalyst Particle Diameter Growth & Density Evolution**





- Regardless of sample size, all batches peaked at around 20 minutes
- M1 diameter is the most stable from 0 to 30 minutes
- In all 4 sample sizes, the particle density decreased over time
- Smallest drops: M1 had a 16% drop and XL had a 12% drop

#### **Conclusion & Next Steps**

- 1. The data suggests that the diameter of the nanotubes peaks at about 20 minutes of reduction time at a temperature of 900°C
- 2. The particle density steadily decreases as reduction time increases in all samples
- 3. Density remains the least unchanged in M1 and XL indicating stability, but M1 has the most stable diameter
- 4. Many samples had a relatively larger standard deviation making statistical significance difficult to determine
- 5. Catalyst is evolving, but exact causes are not fully understood

Next steps:

- 1. Data for 40 minute reduction time can be collected
- 2. Try a higher melting point metal and investigate its thermal stability
- 3. Image the substrate with a Transmission Electron Microscope